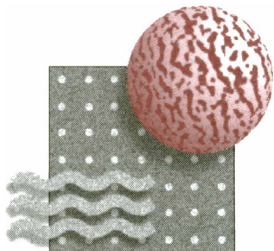


# Carbon Molecular Sieve Membrane as Reactor/Separator for Water Gas Shift Reaction

*DE-FG36-05GO15092*

This presentation does not contain any proprietary, confidential, or otherwise restricted information



**Paul KT Liu**  
**Media and Process Technology Inc.**  
**1155 William Pitt Way**  
**Pittsburgh, PA 15238**  
**Date: May 15, 2007**

**PD7**

# Overview

## ☐ **Project Start Date**

10/1/03

## ☐ **Project End Date**

9/30/07

## ☐ **Percent Complete**

80%

☐ Delivery of 99.999% H<sub>2</sub> with high H<sub>2</sub> recovery ratio

☐ Fabrication of membranes/ module suitable for large scale reactor operation

☐ Demonstration of the membrane reactor process in a significant scale

## ☐ **Total project funding**

- DOE Share: \$1,530,713.
- Contractor Share: \$382,678.

## ☐ **Funding received in FY05 & FY06**

\$300K

## ☐ **Funding received in FY07**

\$200K

☐ No catalyst development activities due to funding limitation

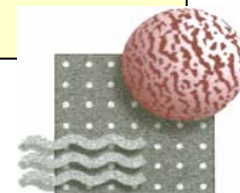
## ☐ **Professor Theo T. Tsotsis**

University of Southern California,  
Catalytic membrane reactor expert

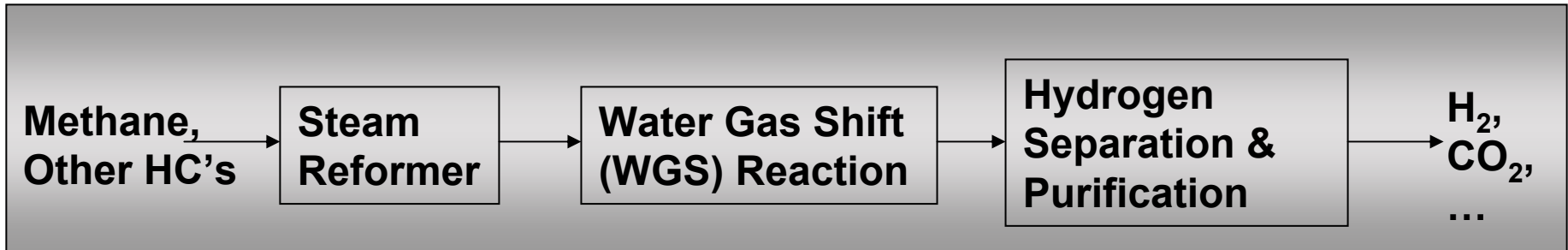
## ☐ **Dr. Babak Fayyaz-Najafi**

Chevron ETC,  
End User Participant

☐ **Dr. Hugh Stitt**, Johnson Matthey,  
Catalyst Manufacturer

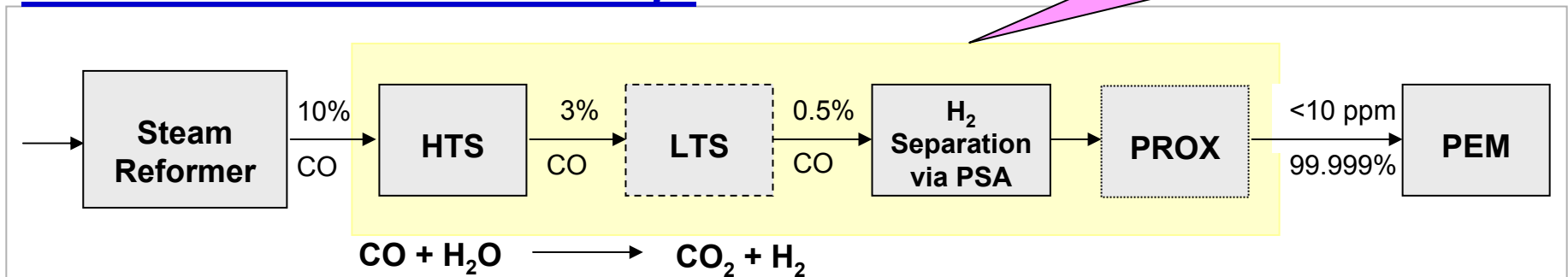


# Hydrogen Production from Steam Reforming

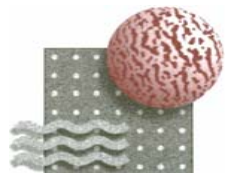


**Our Project Focus**  
*Streamline Unit Operations  
involving CO Conversion/H<sub>2</sub>  
Separation & Purification*

## Conventional Process Concept

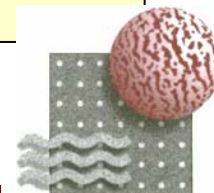
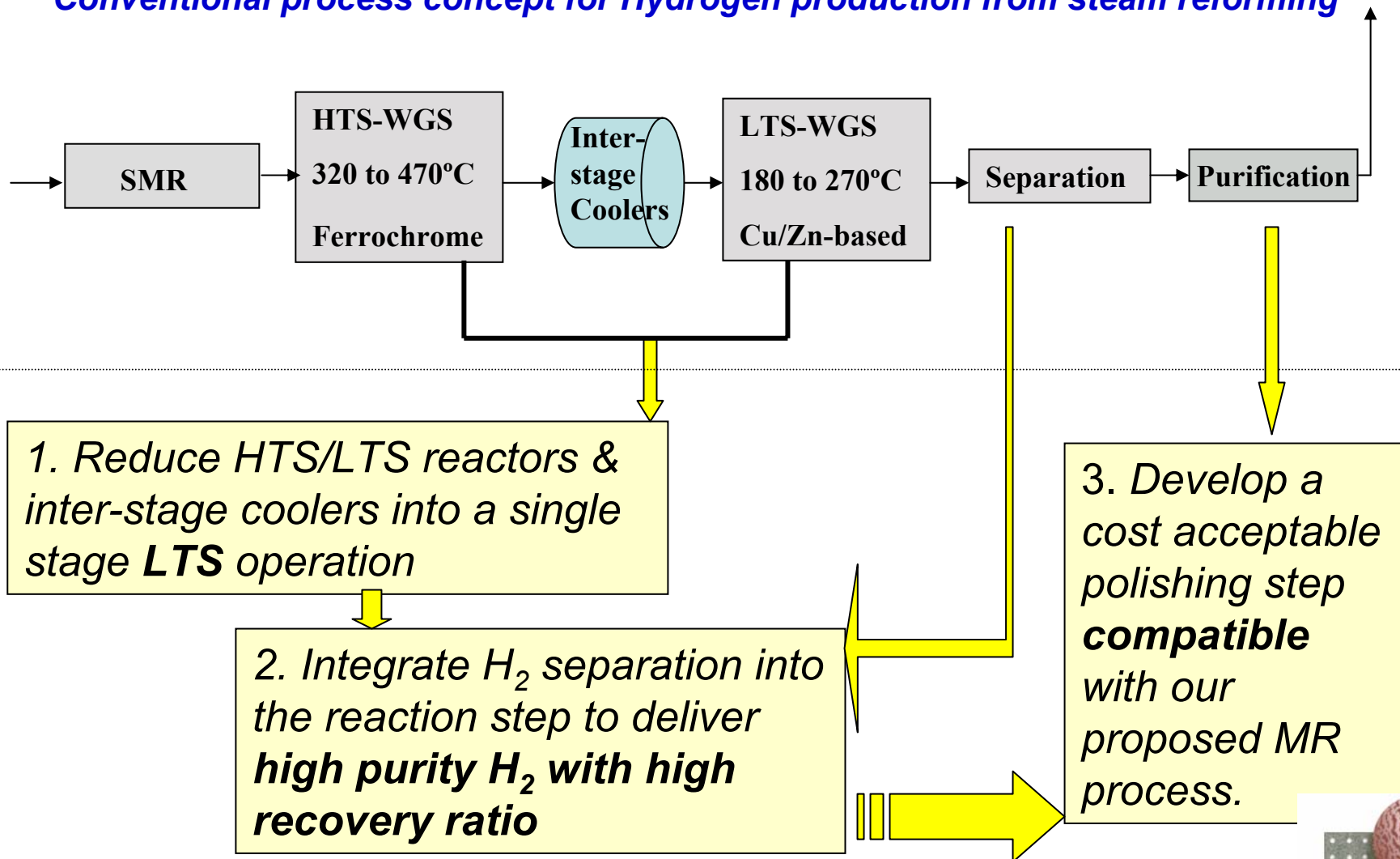


HTS: High Temperature Shift  
LTS: Low Temperature Shift  
PROX: Preferential Oxidation  
PEM: Proton Exchange Membrane Fuel Cell



# OVERALL TECHNICAL STRATEGY

## *Conventional process concept for Hydrogen production from steam reforming*



# Overall Technical Approach

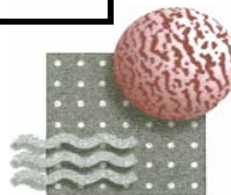
<b>1. Bench-Scale Verification (1<sup>st</sup> to 15<sup>th</sup> month)</b>	<b>2. Pilot Scale Testing (16-24<sup>th</sup> Month)</b>	<b>3. Field Demonstration (25 to 36<sup>th</sup> month)</b>
1.1 Evaluate membrane reactor: use existing membrane & catalyst via math simulation	2.1 Prepare membranes, module, and housing for pilot testing	3.1 Fabricate membranes and membrane reactors and prepare catalysts
1.2 Experimental verification: use upgraded membrane & existing catalyst via bench unit	2.2 Perform pilot scale testing	3.2 Prepare site and install reactor
	2.3 Perform economic analysis & technical evaluation	3.3 Perform field test
1.3 Validate membrane and membrane reactor performance & economics	2.4 Prepare field testing	3.4 Conduct system integration study
		3.5 Finalize economic analysis & refine performance simulation



Technology development team



End user participant



# Technical Approach – Yr II

## ❑ Perform Bench-Top MR Evaluation

- ✓WGS Catalyst evaluation under the proposed operating condition
- ✓Perform MR experiment to verify the prediction of H<sub>2</sub> purity, H<sub>2</sub> recovery and residual CO contaminant by our mathematical model in addition to CO conversion, which was verified in Year I.

## ❑ Tailor Membrane Performance for Proposed MR-based H<sub>2</sub> Production Process.

- ✓Refine existing H<sub>2</sub> selective membranes required for the MR selected for scale-up.
  - hydrogen permeance, and selectivity over CO and CO<sub>2</sub>

## ❑ Perform Process Optimization via Simulation for Economics Analysis

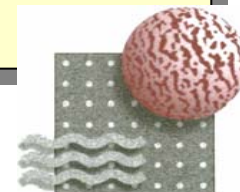
- ✓Identify an optimized MR configuration and operating condition to match the reformer technology developed by our end user participant (Chevron).
- ✓Identify a post treatment configuration to deliver >99.999% hydrogen with <10 ppm CO.

## ❑ Conduct Pilot Test to Verify the Optimized Process

- ✓Using a simulate stream and a full-scale (34”L) single membrane tube for this pilot test.

## ❑ Conduct Technology Validation and Economic Analysis by End User

- ✓Evaluate membrane performance
- ✓Refine mathematical model based upon pilot test results
- ✓Conduct economic analysis



# TECHNICAL ACCOMPLISHMENTS – Yr II

## ❑ Experimental Verification of Mathematical Model

Our MR experimental study has delivered H<sub>2</sub> purity, H<sub>2</sub> recovery ratio, and residual CO contaminant level consistent with prediction by our model. In addition, the effect of reactor temperature has been verified experimentally.

## ❑ Development of A MR-based H<sub>2</sub> Production Process – HiCON

The HiCON process has been developed for the small scale reformer developed by our end user participant (Chevron). A nearly complete CO conversion (i.e., 99+%) can be realistically achieved in contrast to ~70% conversion by HTS and ~95% by HTS + LTS with the conventional reactors.

## ❑ Optimization via Simulation

Process optimization study demonstrates that 97-99% H<sub>2</sub> purity and 98-75% H<sub>2</sub> recovery can be accomplished.

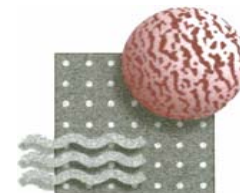
## ❑ Development of A Simple & Cost Effective Polishing Step

Instead of PROX, a simple adsorptive process can be installed as a polishing step for HiCON. Thus, 99.999% H<sub>2</sub> purity and <10 ppm CO can be accomplished. Our preliminary economic analysis indicates nominal cost, e.g., 2-4¢/kg H<sub>2</sub>.

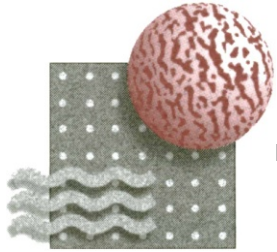
## ❑ Facility & Safety

A barricade has been established ready for performing the proposed HiCON process at a pilot scale.

*In short, we have completed the bench top experimental study and mathematical simulation. The HiCON process has been developed to meet the PEM fuel spec. We are now ready for pilot testing to be performed during the remaining FY2007.*







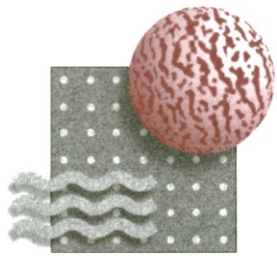
## M&P Ceramic MEMBRANES - Low cost

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*Our Commercial Ceramic Membranes/Bundles and their Substrate*



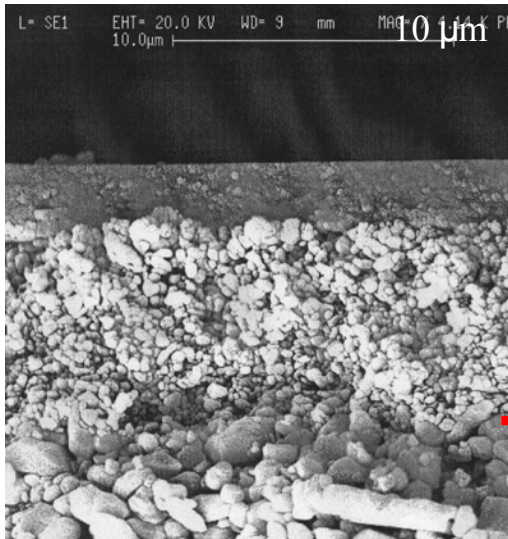




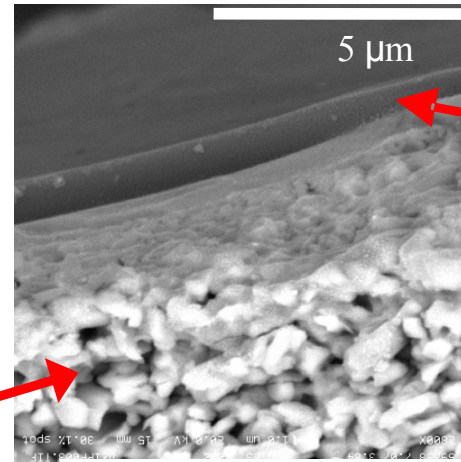
# M&P Emerging Inorganic Membranes

*M&P's Core Technology: Thin film deposition on porous substrates*

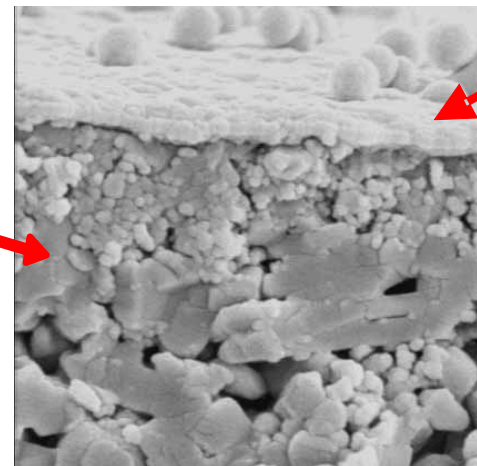
**Inorganic Substrate**



**Ceramic  
Substrate**



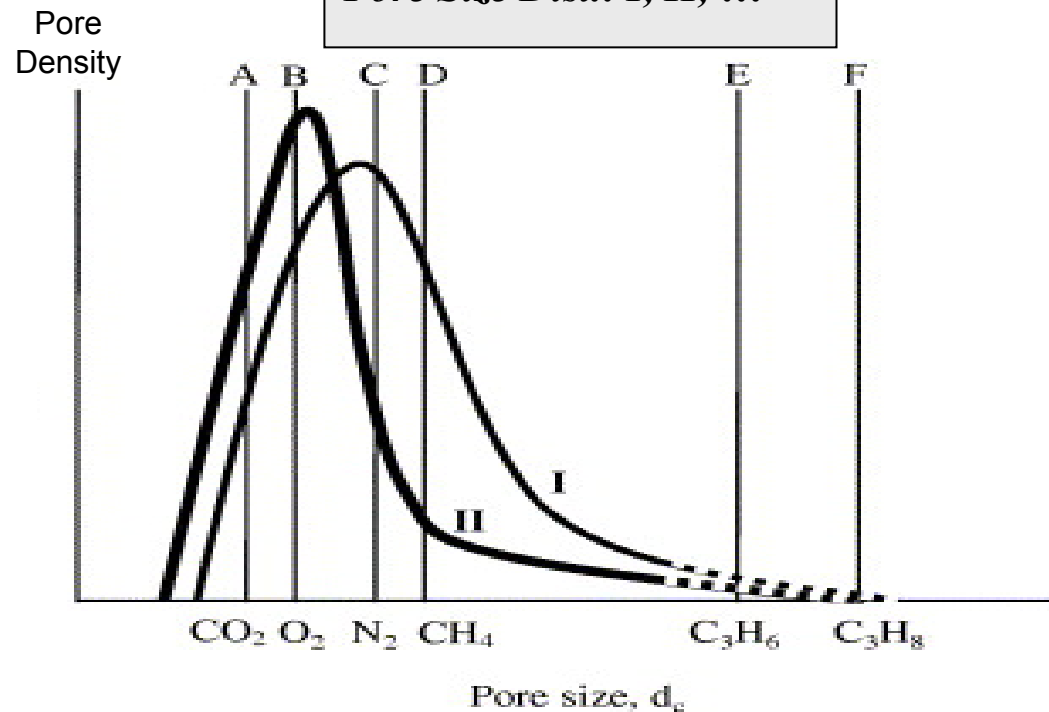
**Carbon  
molecular  
sieve  
(porous)**



**Palladium  
(dense)**

# Carbon Molecular Sieve (CMS) Membranes

*Typical CMS Membrane  
Pore Size Dist.: I, II, ...*

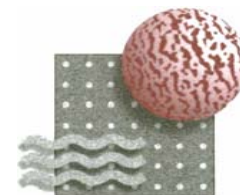


*Unique Features*

**Operating  
Temperature**

**Selectivities**

**Material Stability**



# CMS Membrane Performance Upgrading – Yr II

**Accomplishment: Enhanced  $H_2/CO$  &  $H_2/CO_2$  selectivities without sacrificing  $H_2$  permeance**

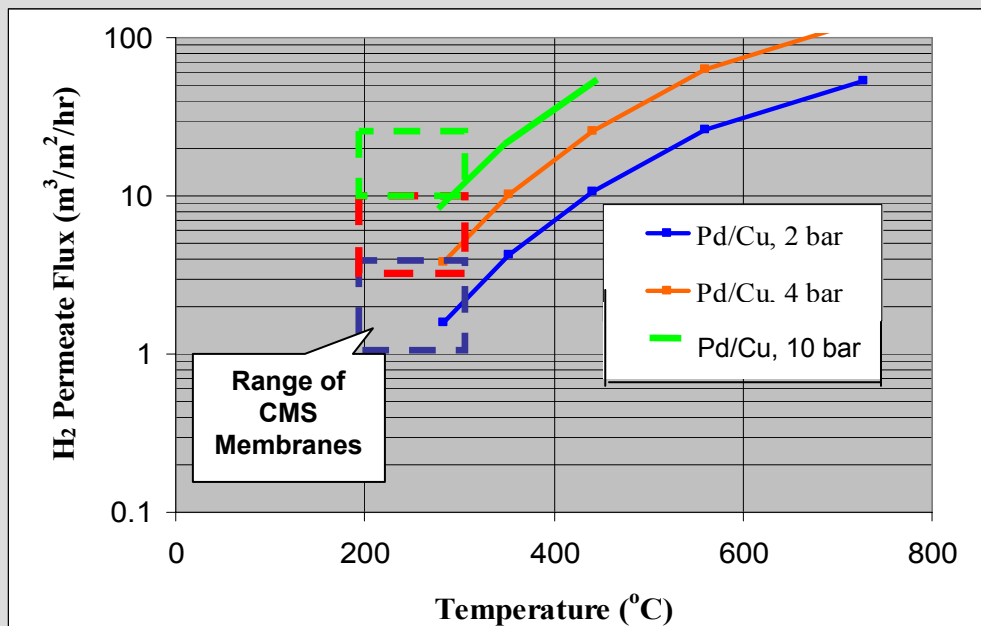
## M&P CMS $H_2$ Selective Membrane

**Typical Hydrogen Permeance:**  
1-3  $m^3/m^2/hr/bar$  at 220°C

## Typical Selectivities:

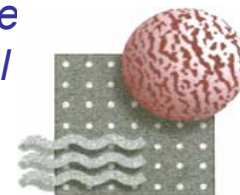
Gas Molecules	Kinetic Diameter [Å]	Selectivity over $N_2$ at 220°C
$H_2$	2.89	1
$CO_2$	3.30	10 to 40
$N_2$	3.64	40 to >80
CO	3.76	50 to >90
$CH_4$	3.80	60 to >100

## CMS vs Pd Alloy Membranes



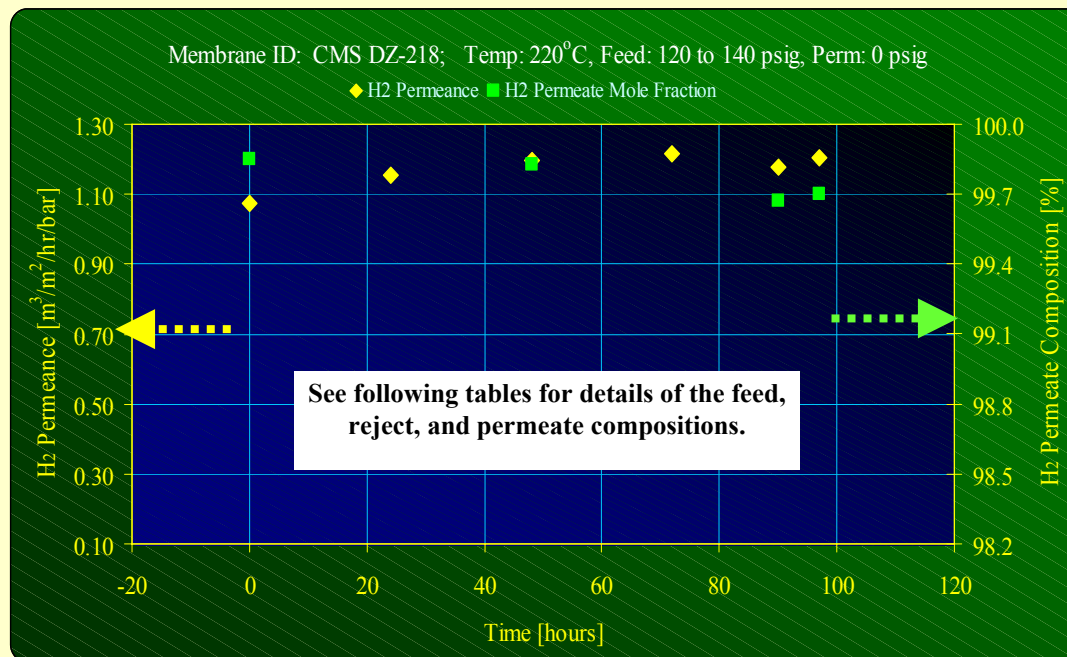
- Assuming 1 micron thickness of Pd/Cu membrane,
- Permeate flux data source: Morreale, B.D., etc, *JMS*, 241(2004) 219
- Feed Pressure as indicated, Permeate Pressure: 1 atm

*During Yr II we have tailored our CMS membrane with the properties above to suite the proposed MR application requirements. Its thermal, hydrothermal and chemical stability under the proposed application environment was demonstrated in Yr I.*



## CMS Membrane: Material Stability at a Pilot Test

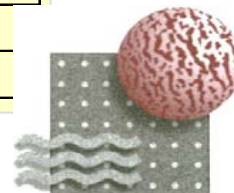
Membrane performance is stable in a 100 hour challenge test conducted at a refinery pilot facility using VGO hydrocracker off-gas in the presence of significant H<sub>2</sub>S, NH<sub>3</sub>, and higher hydrocarbon contamination.



### Gas Stream Compositions, Stage Cut and H<sub>2</sub> Recovery During the VGO Hydrocracker Pilot Test

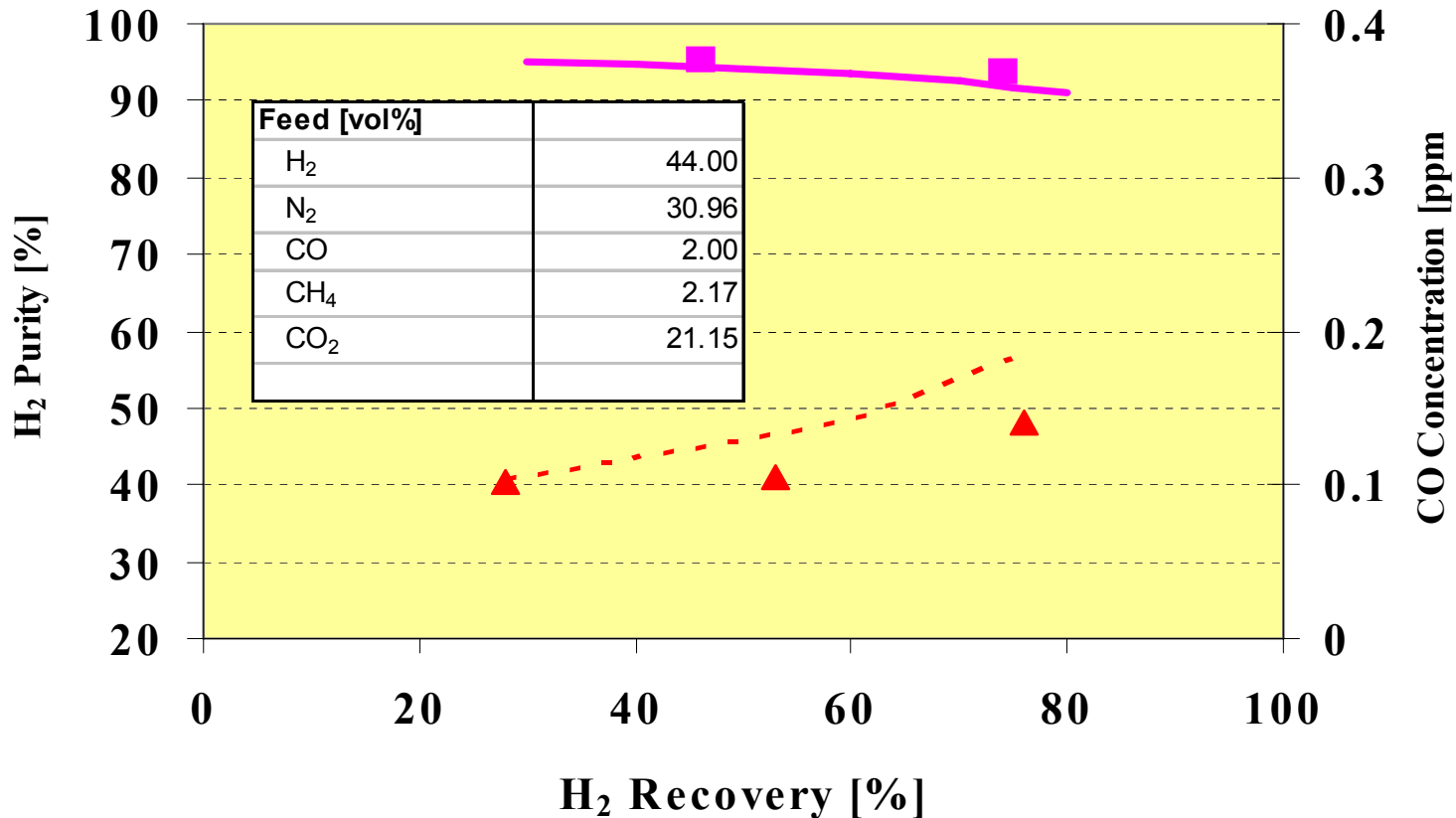
At time = 3 hours				
Gas	Composition [%]			H <sub>2</sub> /Slow Selectivity
	Feed	Reject	Permeate	
H <sub>2</sub> S	5.2	32.0	0.03	163
H <sub>2</sub>	89.9	38.9	99.88	1
C <sub>1</sub>	2.1	12.2	0.08	123
C <sub>2</sub>	0.88	5.4	0.01	~600
C <sub>3</sub> +	1.88	11.6	ND	>1,000
Stage Cut		85%		
H <sub>2</sub> Recovery		92%		

At time = 100 hours				
Gas	Composition [%]			H <sub>2</sub> /Slow Selectivity
	Feed	Reject	Permeate	
H <sub>2</sub> S	4.8	24.5	0.16	74
H <sub>2</sub>	90.8	50.6	99.70	1
C <sub>1</sub>	1.9	9.9	0.06	123
C <sub>2</sub>	0.81	4.2	0.01	~600
C <sub>3</sub> +	1.66	10.7	ND	>1,000
Stage Cut		80%		
H <sub>2</sub> Recovery		85%		

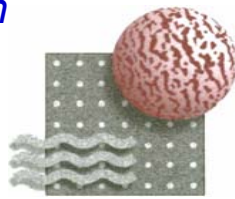


# Experimental Verification:

## Mixture Separation vis CMS Membranes



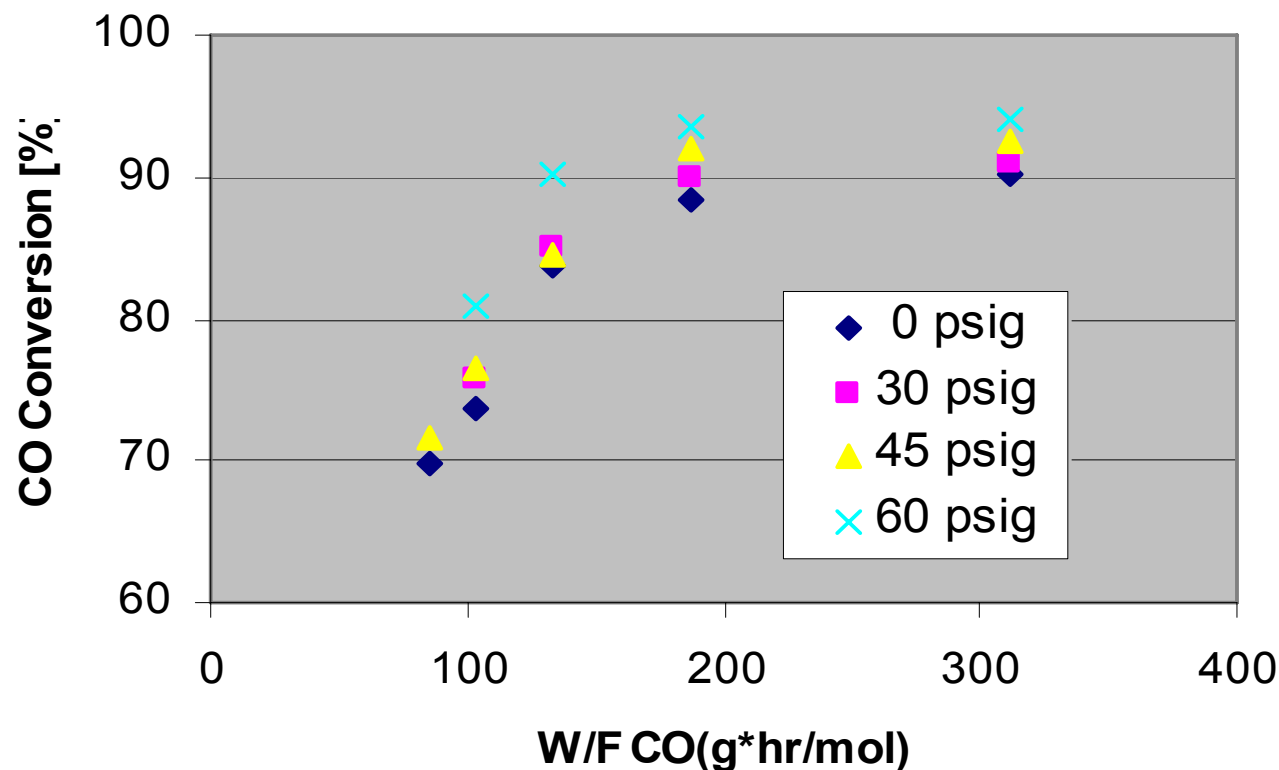
*The performance of our CMS membrane was demonstrated in mixture separation using a synthetic reformate shown above. Further our mathematical model can reliably predict the permeate composition vs H<sub>2</sub> recovery.*



# Water Gas Shift Reaction Kinetic Study

Cu/Zn catalyst: CO Conversion vs Pressure

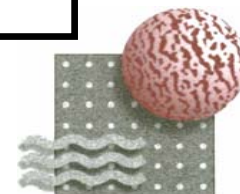
$T = 250^{\circ}\text{C}$  ,  $P_{\text{feed}} = 1 \text{ to } 5 \text{ atm}$ , Feed Composition:  $\text{H}_2:\text{H}_2\text{O}:\text{CO}:\text{CO}_2 = 5:3:1:0.5$



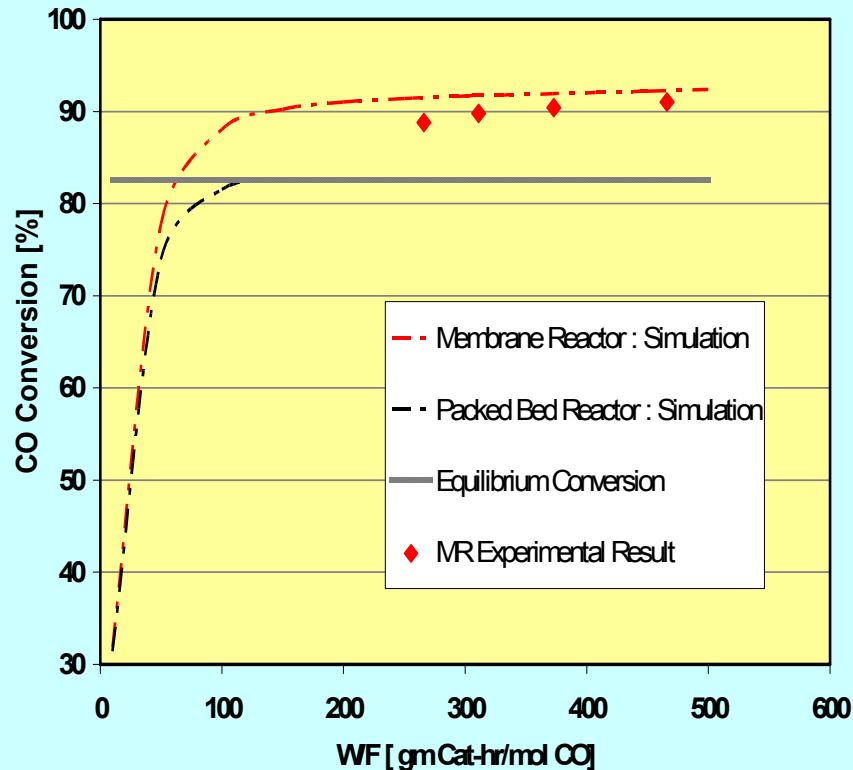
$$\text{Rate} = k_0 \cdot (1 - \text{Beta}) \cdot (P_{\text{CO}}) \cdot (P_{\text{H}_2\text{O}})^{1.8} (P_{\text{CO}_2})^{-1.4} (P_{\text{H}_2})^{-0.5} P_t^{-0.8}$$

*Rate constant suitable for the proposed application environment was determined, which has been used for our simulation study.*

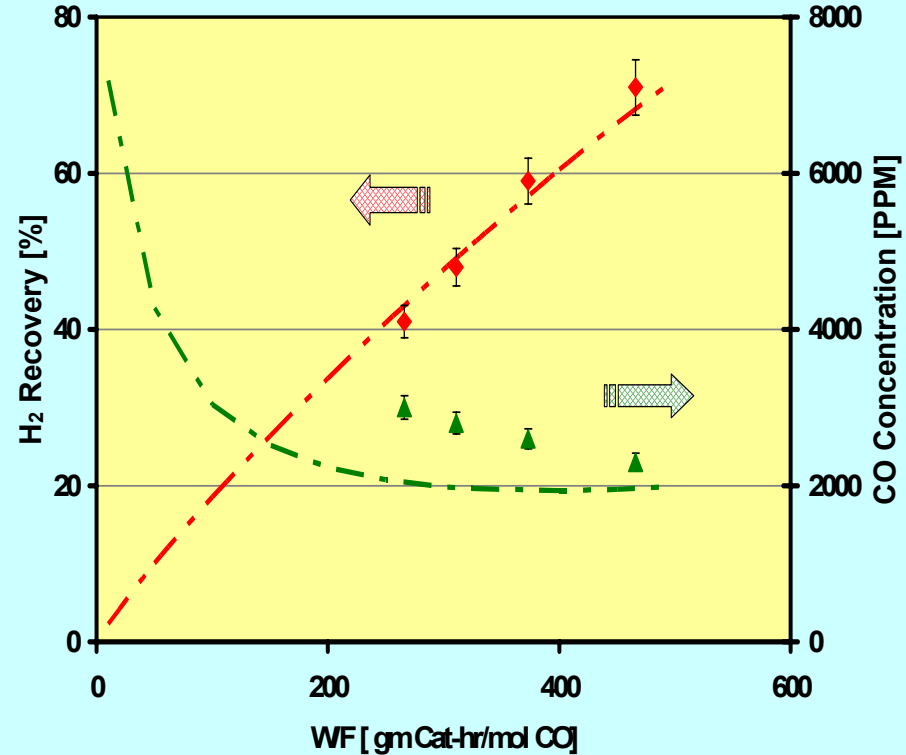
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# Bench Top Membrane Reactor Study: Experimental vs Simulated



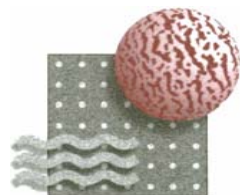
**CO Conversion vs W/F**



**H<sub>2</sub> Recovery, CO Concentration vs W/F**

W/F: Ratio of Catalyst Dosage to Feed FlowRate

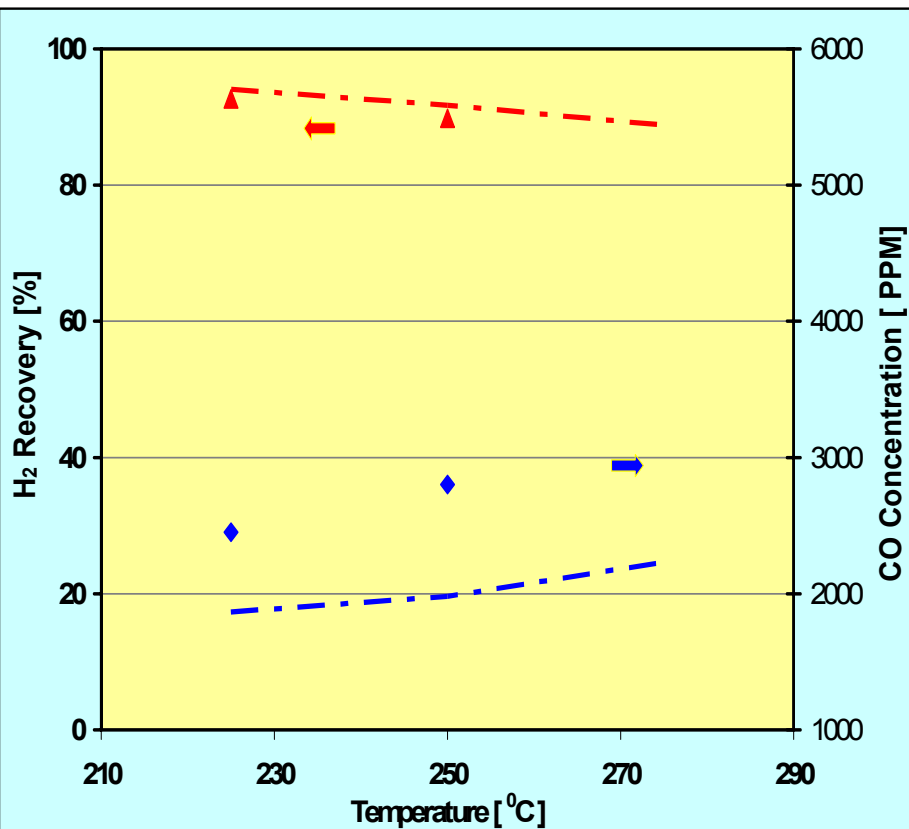
*Our prediction on CO conversion, H<sub>2</sub> recovery and CO residual level was verified with experimental results obtained from our bench top membrane reactor.*



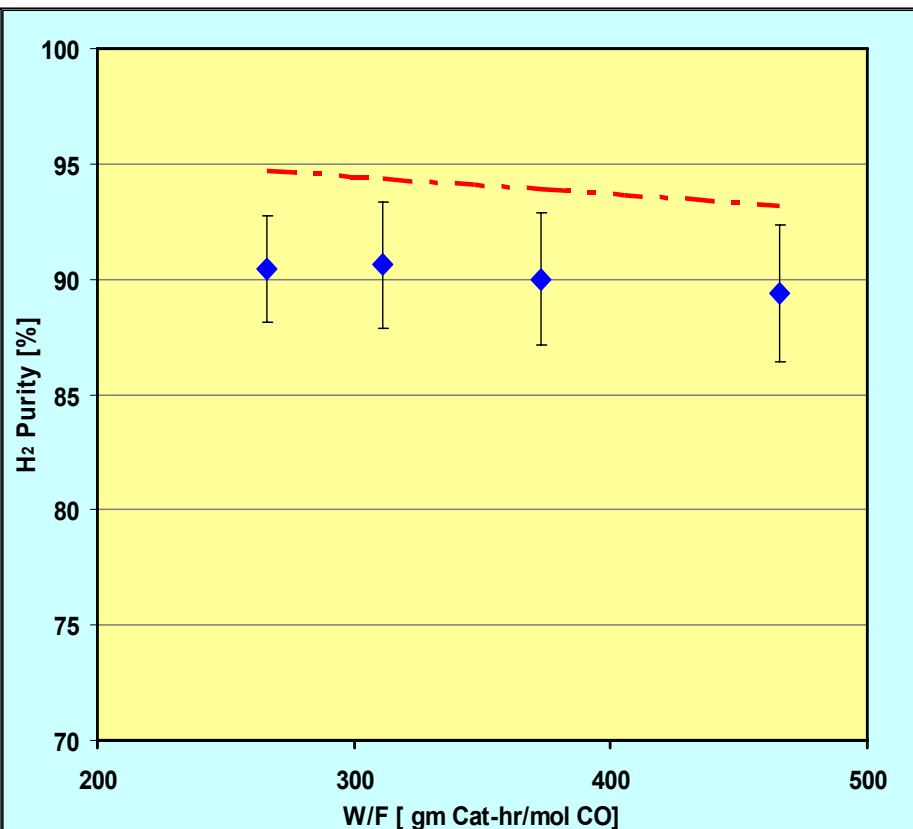


# Bench Top Membrane Reactor Study

## Experimental vs Simulated & Verification of Mathematical Model

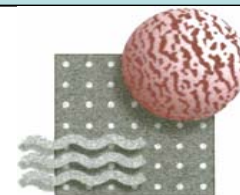


**H<sub>2</sub> Recovery & CO Conc. vs W/F**



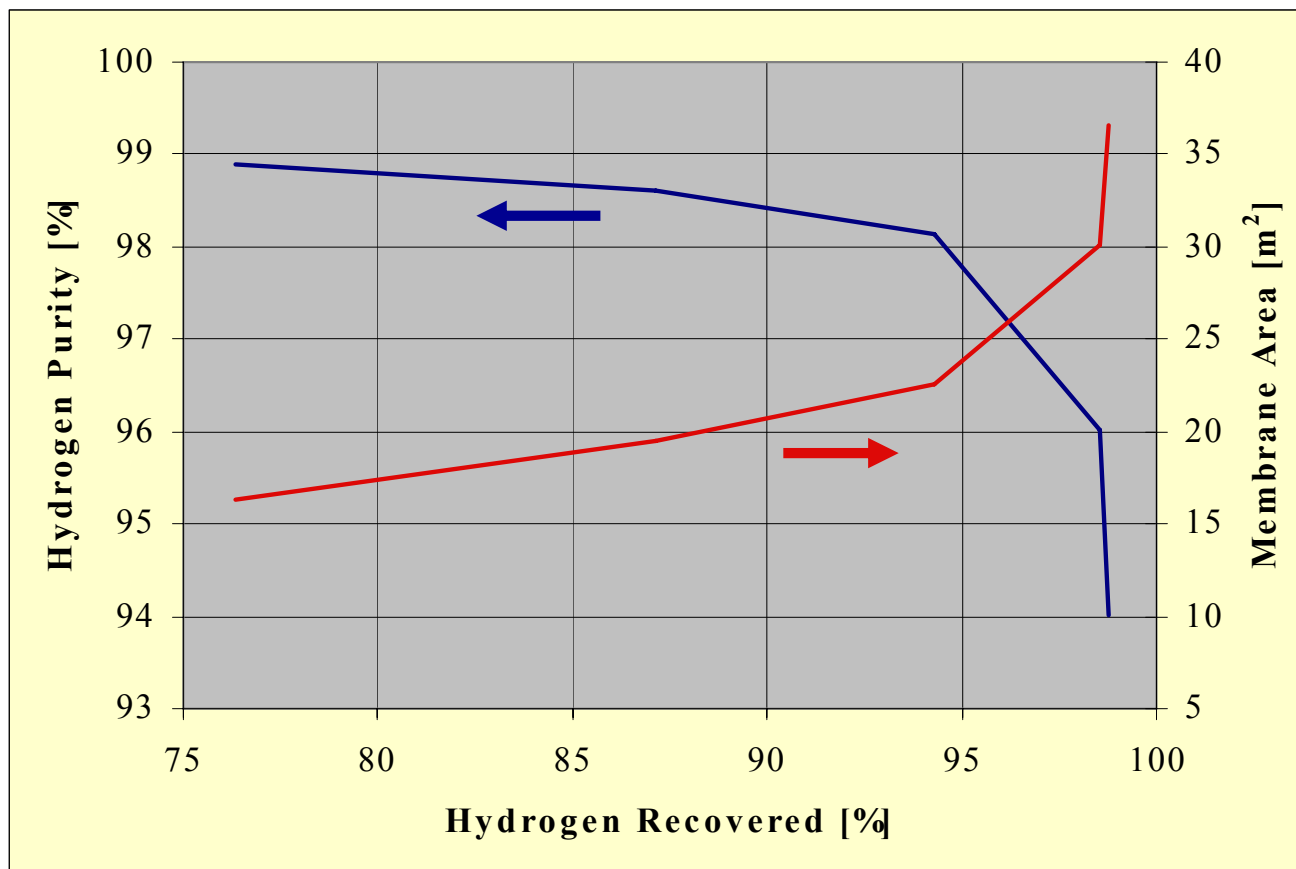
**H<sub>2</sub> Purity vs W/F**

*Our prediction on H<sub>2</sub> purity and effect of temperature was verified with experimental results obtained from our bench top membrane reactor.*



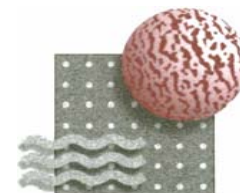
# Performance of Our Proposed Membrane Reactor Process

via Mathematical Simulation, basis: 100 kg/day  $H_2$  feed



*Our proposed HiCON can deliver 96-99%  $H_2$  purity with 98-77%  $H_2$  recovery with a modest membrane surface area requirement.*

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# Preliminary Economic Analysis: Post Treatment Capital and Operating Cost

Target: 99.999% purity H<sub>2</sub>

Basis: 1500 kg/day H<sub>2</sub> production

## Case A: Temperature Swing Adsorption (TSA) Integrated with Membrane Reactor

Adsorption temperature [C]				50
Pressurization cycle [min]				5
Adsorption Cycle [min]				175
Temperature Swing Regeneration [min]				180
Feed Purity [%]	99	97	93	
Adsorber ID [in]	12.6	15.7	19.8	
Adsorber Height [ft]	11.8	19.9	29	
Capital Cost* [\$]	134,598	214,249	333,304	
Capital Recovery Cost [¢/Kg H <sub>2</sub> ]	4.1	6.5	10.1	
Hydrogen Yield [%]	~100	~100	~100	

**Bulk  
Hydrogen  
Cost**

**at Production  
Point  
via Methane  
Steam  
Reforming**

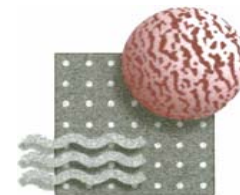
**\$1 – 2.4/Kg H<sub>2</sub>  
for 22-600  
tons/day with  
\$3.5-7/GJ NG**

\* Example of Capital Cost Estimate: for 99% purity case

For Quantity of 4 Adsorbers	
Purchase Price of Pres Vessels, fob	\$42,032
Purchase Price of Zeolite, fob	\$2,162
Purchase Price of Support, fob	\$39
Delivery	\$2,212
Installation	\$51,090
Purchased, Deliverd & Installed	\$97,535
Piping, Valving & Instumentation	\$19,507
Total Fixed Capital Investment	\$117,042
Other One-Time Costs	\$17,556
Other One-Time Costs	\$134,598
Not including heating equipment for TSA.	

Our preliminary analysis indicates that the incremental cost for the developed post treatment scheme is very insignificant

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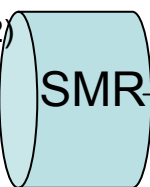
# Distributed Hydrogen Production Process

## HiCON

Schemes Unit Operations	Scheme #1		Scheme #2		Scheme #3	
	H <sub>2</sub> Purity [%]	H <sub>2</sub> Recovery Ratio [%]	H <sub>2</sub> Purity [%]	H <sub>2</sub> Recovery Ratio [%]	H <sub>2</sub> Purity [%]	H <sub>2</sub> Recovery Ratio [%]
Membrane Reactor/Separator (HiCON)	98.5	90	97.0	97	93.0	98
Polishing Step	99.999	100	99.999	100	99.999	100
Cumulative Performance	99.999	90	99.999	97	99.999	98

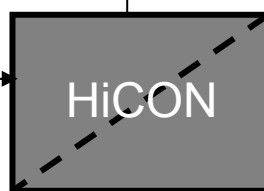
### Feed Ratio

CH<sub>4</sub>: 24%(1)  
H<sub>2</sub>: 5%(0.2)  
CO: 0  
CO<sub>2</sub>: 0  
H<sub>2</sub>O: 71%(3)  
800°C  
10 bar



CH<sub>4</sub>: 2.5%  
H<sub>2</sub>: 52.3%  
CO: 8.9%  
CO<sub>2</sub>: 5.6%  
H<sub>2</sub>O: 30.8%  
10 kmole/hr  
800°C  
10 bar

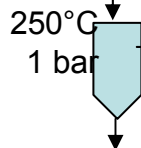
250°C  
10 bar  
Heat  
Exchanger



CMS  
Membrane  
Reactor

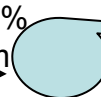
CH<sub>4</sub>: 10%  
H<sub>2</sub>: 26%  
CO<sub>2</sub>: 55%  
H<sub>2</sub>O: 9%  
CO: trace  
2.4 kmole/hr  
250°C

H<sub>2</sub>: 73%  
H<sub>2</sub>O: 26%  
CO<sub>2</sub> & CH<sub>4</sub>: 1%  
CO: ~20 ppm  
7.6 kmole/hr  
250°C  
1 bar



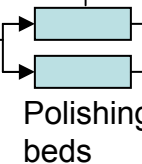
condensate

H<sub>2</sub>: 98%  
CO<sub>2</sub> & CH<sub>4</sub>: 2%  
CO: ~30 ppm  
5.6 kmol/hr  
30°C  
1 bar



compressor

Purge: H<sub>2</sub> & CO



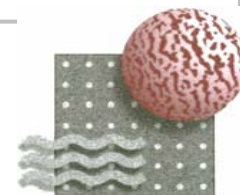
### To Storage

H<sub>2</sub>: 99.999%  
CO: <10 ppm  
~5.6 kmole/hr  
30°C  
30 bar

Not all heating/cooling  
requirements are  
shown

It appears that our HiCON process is able to deliver nearly 90% hydrogen recovery with 99+% purity according to our simulation.

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# BENCHMARKING: EXISTING PSA/PROX

*Product Stream from Our Enduser Reformer : 10 bar, 10-3% CO*

## Our HiCON Process

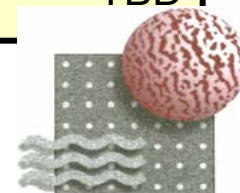
CH <sub>4</sub> : 1 H <sub>2</sub> : 0.2 CO: 0 CO <sub>2</sub> : 0 H <sub>2</sub> O: 3 800°C 10 bar		CH <sub>4</sub> : 2.5% H <sub>2</sub> : 52.3% CO: 8.9% CO <sub>2</sub> : 5.6% H <sub>2</sub> O: 30.8% 800°C 10 bar		H <sub>2</sub> : 98% CO <sub>2</sub> &CH <sub>4</sub> :2% CO: ~30 ppm 30°C 1 bar		99.999% H <sub>2</sub> <10 ppm CO
	<b>SMR</b>		<b>Membrane Reactor</b>		<b>Post Treatment</b>	
	<b>Small scale</b>		<b>WGS (LTS)</b>		<b>Simple Adsorption</b>	
			<b>H<sub>2</sub> Separation</b>			

## Benchmarking

Performance Criteria	Conventional/PSA	OUR HiCON
CO Conversopm [%]	75 - 95	99
H <sub>2</sub> Recovery [%]	70 - 85	~90
Product, H <sub>2</sub> , pressure [psi]	150	15
Reject, CO <sub>2</sub> , pressure [psi]	15	150
Capital Cost [\$]	TBD	TBD

*Economic analysis is under preparation currently, and will be presented in the meeting.*

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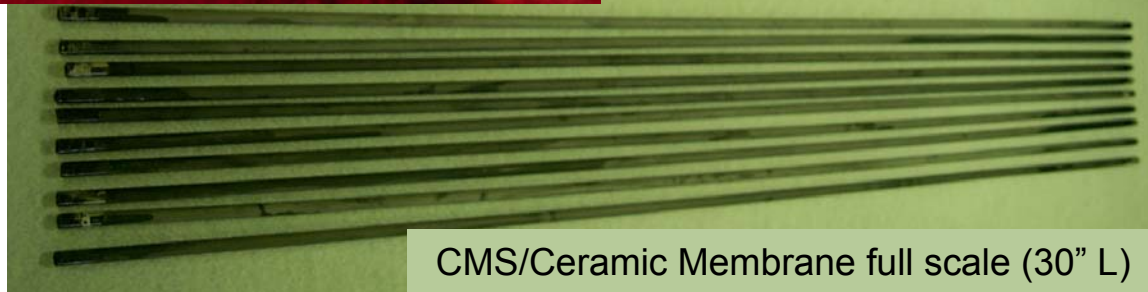


# MEMBRANES, BUNDLE AND MODULE PREPARATION

Pilot Scale Module of CMS/ceramic Membrane (1.5" diameter and 30"L)



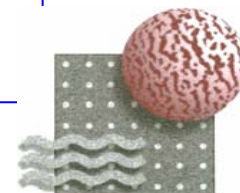
← Our full-scale ceramic membrane module (3 - 4" dia, prototype) for gas applications



CMS/Ceramic Membrane full scale (30" L)

These membranes and modules were adapted from our existing commercial ceramic membrane products and modules.

- The pilot scale module is currently under a pilot test for hydrogen recovery.
- The full-scale single tube (30"L) will be used for our pilot scale MR test in FY2007.





# M&P CMS H<sub>2</sub> SELECTIVE MEMBRANES – PILOT TEST

Engineering Demonstration Facility, Startech Environmental Corp.



2 ton/day Plasma Conversion System  
(based upon MSW)

200 SCFM PCG  
(plasma converted gas)



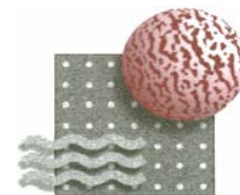
## Performance Results

50/50 H<sub>2</sub>/CO at 18 SCFM and ~100°C

87/13 H<sub>2</sub>/CO at 82% H<sub>2</sub> Recovery for Stage 1 at 9.5 barg

>99% H<sub>2</sub> at 93% H<sub>2</sub> Recovery for Stage 2 at 6.8 barg

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# MEMBRANE REACTOR: PILOT SCALE TESTING FACILITY

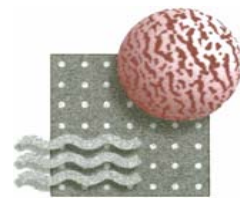
for a Full-Scale (34"L) Single Tube Membrane Reactor



We have designed and constructed this barricade to perform the membrane reactor study involving high temperature and high pressure CO & H<sub>2</sub>. The single full-scale membrane tube (34"L) as a reactor is housed within this barricade. Its temperature is controlled in-situ by an electric tube furnace. Unique safety features include:

- The barricade can be constantly purged with inert gas. Any leak in H<sub>2</sub> and CO can be detected via the purge gas analysis.
- The barricade has a water leg to allow the surge under the worst case scenario: explosion.

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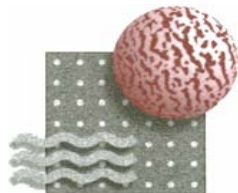
# Future Work

## Remainder of FY 2007

- ☐ Complete pilot scale testing using a single, full-scale hydrogen selective membrane and synthetic feed to demonstrate the optimized HiCON process.
- ☐ Complete the preliminary economic analysis for hydrogen production via the developed HiCON process by our end user.

## FY 2008 and Beyond

- ☐ Depending upon the budget availability, the field demonstration with a pilot scale unit as originally planned will be pursued.



# SUMMARY

## Our Project Team Mission

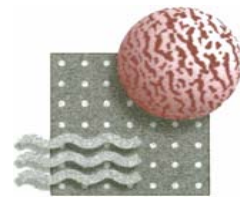
Our project team composed of a membrane manufacturer, a catalyst manufacturer, an end user and an academic institute is well positioned to overcome the commercialization barriers associated with the membrane reactor while the distributed hydrogen production is an ideal platform to showcase the MR technology.

## Our Accomplishments

- We have completed the bench top experimental study and mathematical simulation to demonstrate our HiCON process to deliver **99+%** CO conversion with **97-99%** purity and **98-75%** H<sub>2</sub> recovery via a simple MR process, uniquely suitable for the distributed hydrogen production.
- Although membranes are not ideal to deliver **99.999% purity with trace CO contaminant**. Our study indicates that a cost acceptable post treatment unique to our proposed process can achieve this target. This in conjunction with our HiCON process offers a practical and economically viable process to meet the stringent feed quality requirement for PEM.
- We have established a **pilot scale testing facility** for performing a pilot scale test to verify the optimized HiCON process using a full-scale membrane tube with synthetic feed, which is expected to be completed by the end of FY 2007.

*In short, with the budget available, we anticipate to complete the minimum tasks required to take this HiCON process to the next step for field demonstration.*

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# ACKNOWLEDGEMENT

## US DOE Project Managers

- Arlene Anderson
- Carolyn Elam

## Our Project Team Members

- Theo T. Tsotsis, University of Southern California
- Babak Fayyaz-Najafi & John Wind, Chevron ETC
- Hugh Stitt, Johnson Matthey
- Richard J. Ciora, Jr. Media and Process Tech Inc.

